

Apparatus and process for removing surface regions of a component

The invention relates to an apparatus and a process for
5 removing surface regions of a component as described in
claim 1 or 2, respectively.

Hitherto, components which have been coated with
10 coatings of type MCrAlY or ZrO₂ have had the coating
removed, for example, by acid stripping in combination
with sand blasting or by high-pressure water blasting.

EP 1 122 323 A1 and US 5,944,909 show examples of the
chemical removal of surface regions.

15 EP 1 941 34 A1, EP 1 010 782 A1 and US-A 6,165,345
disclose methods for the electrochemical removal of
metallic coatings (stripping).

20 The processes listed above are time-consuming and
therefore expensive.

It is an object of the invention to provide an
apparatus and a process in which the removal of the
25 coating takes place more quickly and economically.

The object is achieved by an apparatus and a process
for the removal of surface regions from a component as
described in claims 1 and 2, respectively.

30 Further advantageous configurations and process steps
are listed in the corresponding subclaims.

In the drawing:

Figure 1 shows an apparatus according to the invention,
Figure 2 shows a time curve of a current of a current
pulse generator, and
5 Figure 3 shows a further time curve of a current from a
current pulse generator.

Figure 1 shows an apparatus 1 according to the invention. The apparatus 1 comprises a vessel 4 in
10 which an electrolyte 7 there is arranged. An electrode
10 and a component 13 are arranged in the electrolyte
7. The electrode 10 and the component 13 are
electrically connected to a current/voltage pulse
generator 16. The component 13 is, for example, a
15 coated turbine blade or vane, the substrate of which is
a nickel- or cobalt- base superalloy, to which a
metallic layer has been applied to serve, for example,
as a corrosion-resistant or anchoring layer. A layer of
this type in particular has the composition MCrAlY,
20 where M stands for an element iron, cobalt or nickel.

The coating has been corroded during use of the turbine
blade or vane 13. The surface region 25 which has been
formed as a result (as indicated by dashed lines) is to
25 be removed by the process according to the invention
and the apparatus 1 according to the invention. It is
also possible for layer regions 25 which have been
formed by corrosion, oxidation or other forms of
degradation to be removed from a component 13 which
30 does not have a coating, these layer regions being in
the vicinity of the surface.

The current pulse generator 16 generates a pulsed
current/voltage signal (Figure 2).

35 An ultrasound probe 19, which is operated by an
ultrasound source 22, may optionally be arranged in the
electrolyte 7.

The ultrasound excitation improves the hydrodynamics of the process and thereby assists the electrochemical reaction.

5 Figure 2 shows an example of a current/voltage curve of the current/voltage pulse generator 16.

The current pulse signal or the voltage pulse is, for example, square-wave (pulse shape) and has a pulse 10 duration t_{on} . Between the individual pulses there is an interval of length t_{off} . Furthermore, the current pulse signal is defined by its current level I_{max} .

15 The current (I_{max}) which flows between the electrode 10 and the component 13, the pulse duration (t_{on}) and the pulse interval (t_{off}) have a significant influence on 20 the electrochemical reaction by accelerating the latter.

Figure 3 shows an example of a series of current pulses 20 40 which are repeated. A sequence 34 comprises at least two blocks 77. Each block 77 comprises at least one current pulse 40. A current pulse 40 is characterized by its duration t_{on} , the level I_{max} and its pulse shape (square-wave, delta, etc.). Other important process 25 parameters are the intervals between the individual current pulses 40 (t_{off}) and the intervals between the blocks 77.

The sequence 34 comprises, for example, a first block 30 77 of three current pulses 40 between each of which there is an interval. This is followed by a second block 77, which has a higher current level and comprises six current pulses 40. After a further interval, there then follow four current pulses 40 in 35 the opposite direction, i.e. with a reversed polarity.

The sequence 34 is finished by a further block 77 of four current pulses.

The sequence 34 can be repeated a number of times.

5 The individual pulse times t_{on} are preferably of the order of magnitude of approximately 1 to 10 milliseconds. The time duration of the block 77 is of the order of magnitude of up to 10 seconds, so that up to 500 pulses are emitted in one block 77.

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The application of a low potential (base current) both during the pulse sequences and during the intervals is optionally possible.

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The parameters of a block 77 are matched to a constituent of an alloy which, by way of example, is to be removed in order to optimize the removal of this constituent. This can be determined in individual tests.